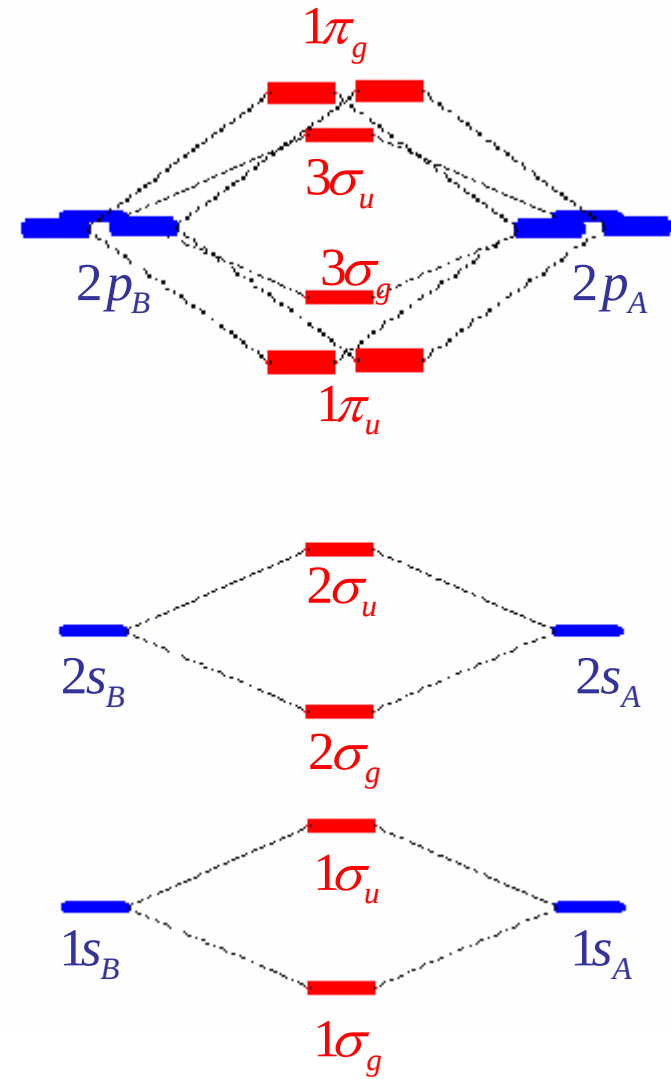
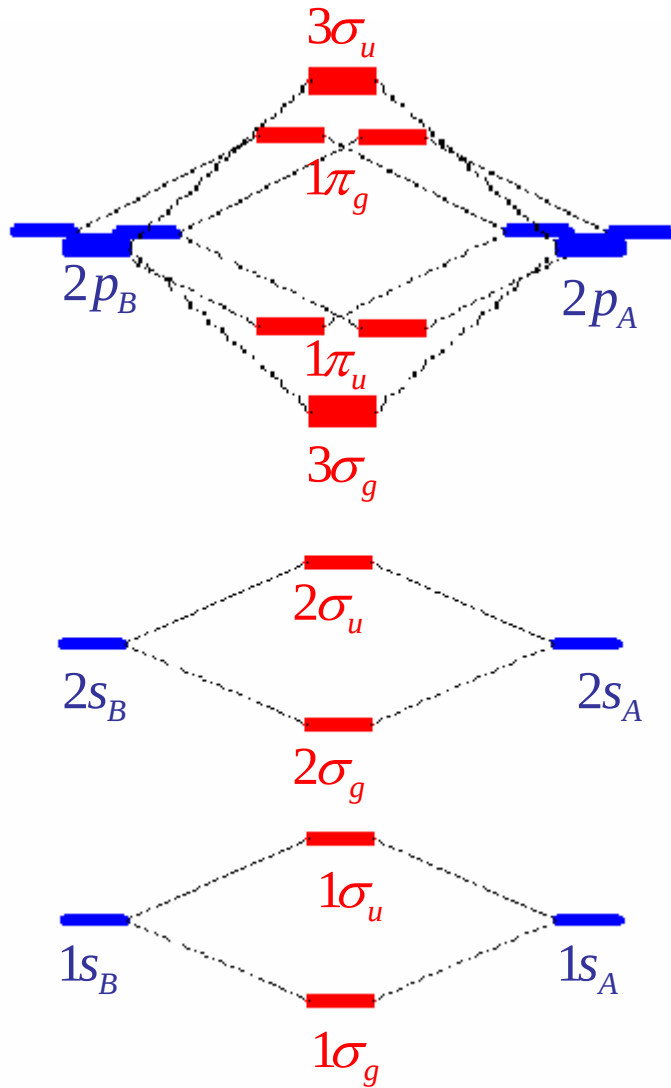


Molécules X_2

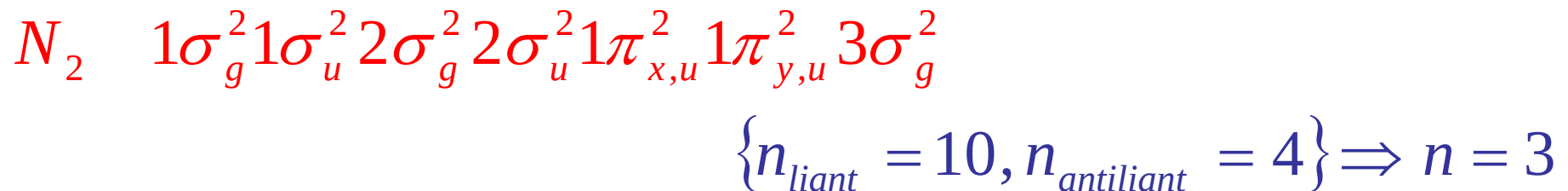


Molécules X_2

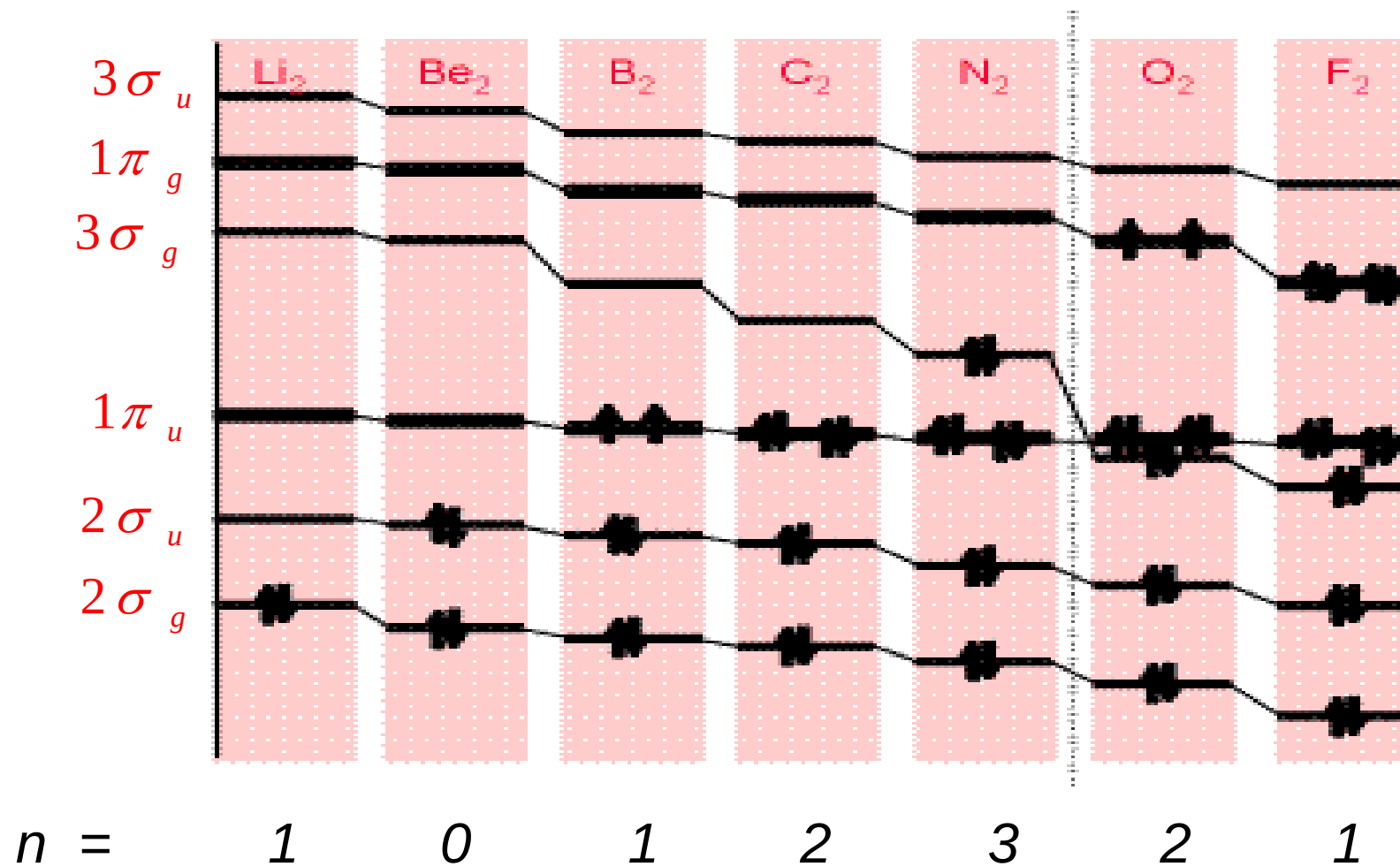
- Ordre de liaison: (ou indice de liaison)

$$n = \frac{1}{2}(n_{\text{liant}} - n_{\text{antiliant}})$$

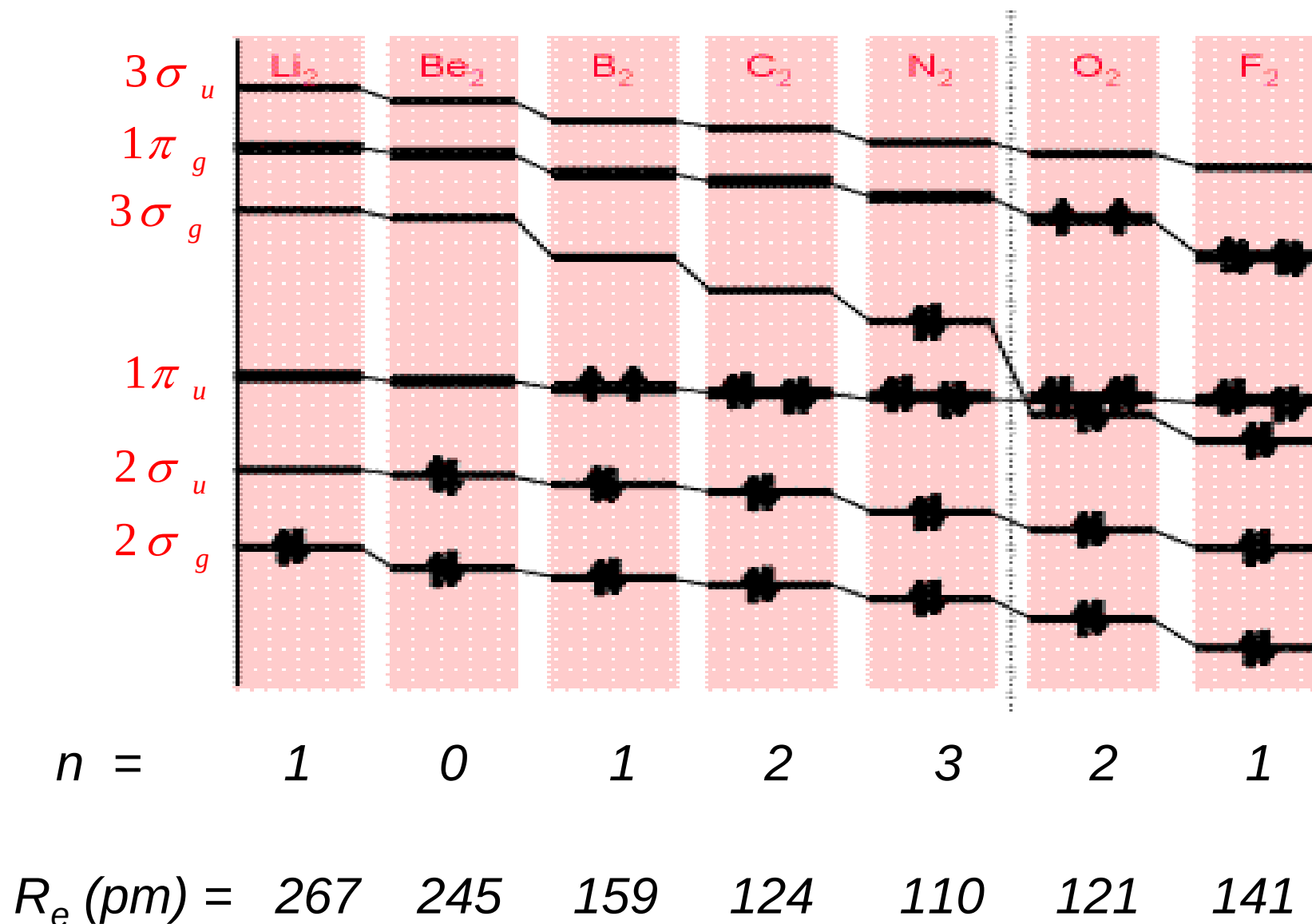
Exemples:



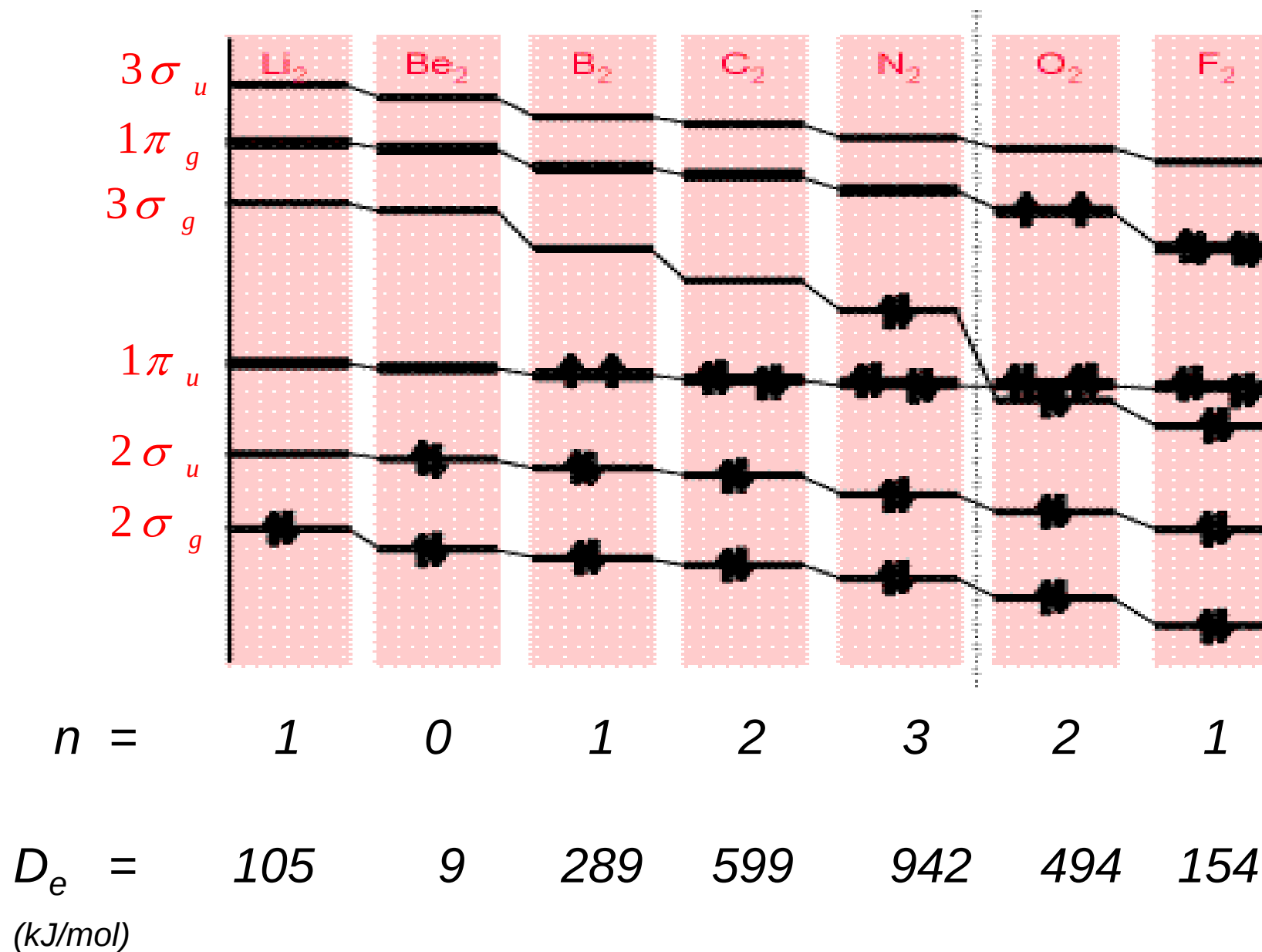
Molécules X_2



Molécules X_2



Molécules X_2



Molécules AB faiblement polaires

SI

$$e_A \cong e_B$$

Molécules AB faiblement polaires

SI

$$e_A \cong e_B$$

Nomenclature des OM



A ₂	AB
1σ _g	1σ
1σ _u	1σ*
2σ _g	2σ
2σ _u	2σ*
3σ _g	3σ
3σ _u	3σ*
1π _{u,x}	1π _x
1π _{g,x}	1π _x *
1π _{u,y}	1π _y
1π _{g,y}	1π _y *

Molécules AB faiblement polaires

SI

$$e_A \cong e_B$$

Nomenclature des OM



A ₂	AB	caractère
1σ _g	1σ	liante
1σ _u	1σ*	antiliante
2σ _g	2σ	liante
2σ _u	2σ*	antiliante
3σ _g	3σ	liante
3σ _u	3σ*	antiliante
1π _{u,x}	1π _x	liante
1π _{g,x}	1π _x *	antiliante
1π _{u,y}	1π _y	liante
1π _{g,y}	1π _y *	antiliante

Molécules AB faiblement polaires

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Nomenclature des OM



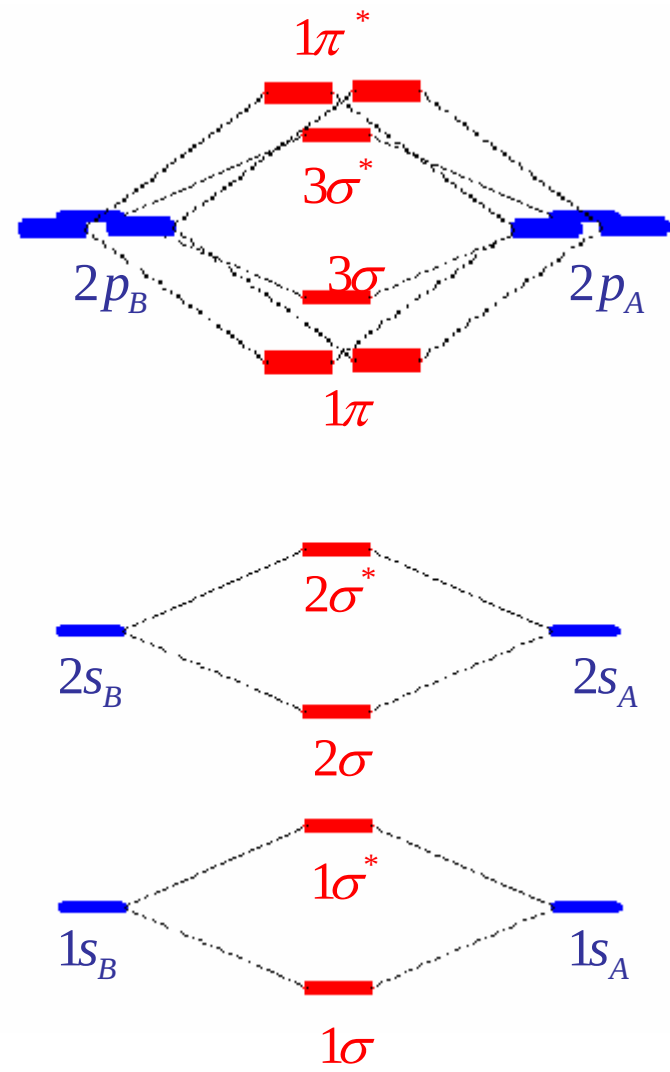
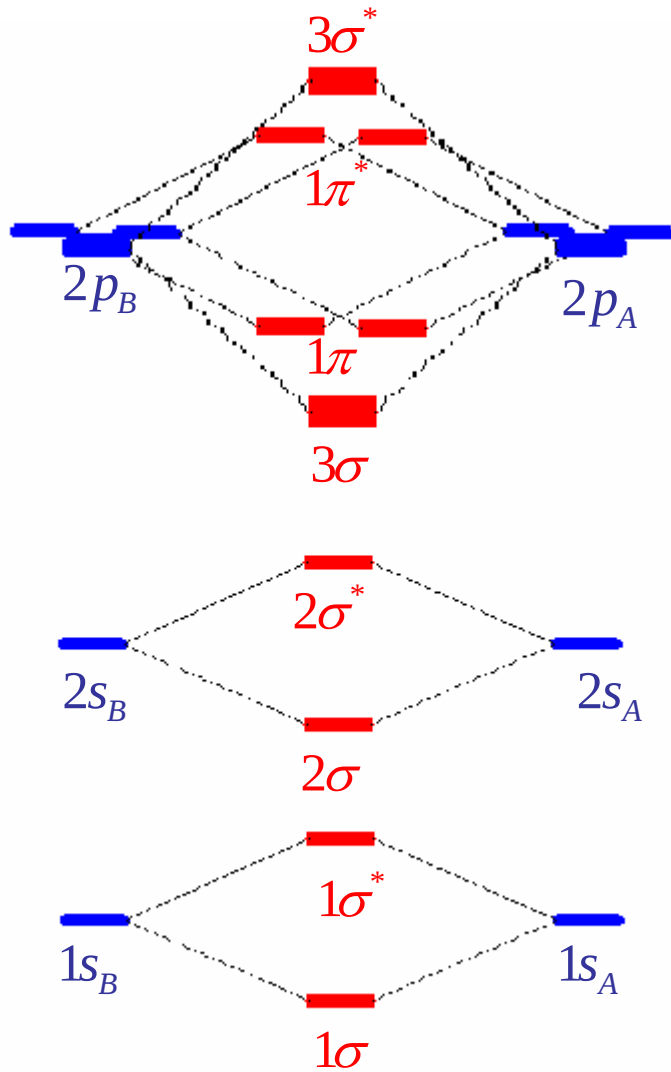
Ordre de liaison:

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toujours

A ₂	AB	caractère
1σ _g	1σ	liante
1σ _u	1σ*	antiliante
2σ _g	2σ	liante
2σ _u	2σ*	antiliante
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3σ _u	3σ*	antiliante
1π _{u,x}	1π _x	liante
1π _{g,x}	1π _x *	antiliante
1π _{u,y}	1π _y	liante
1π _{g,y}	1π _y *	antiliante

Molécules AB faiblement polaires



Molécules AB (fortement) polaires

SI

$$e_A \prec e_B$$

Molécules AB (fortement) polaires

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$$e_A \prec e_B$$

- considérer cas par cas

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- respecter règles LCAO

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- respecter règles LCAO
- noter généralement:
 - l'existence d'orbitales non-liantes, dites **OM de cœur**

Molécules AB (fortement) polaires

SI

$$e_A \prec e_B$$

- considérer cas par cas
- respecter règles LCAO
- noter généralement:
 - l'existence d'orbitales non-liantes, dites **OM de cœur**
 - des **interactions d'OA de valence** seulement

Molécules AB (fortement) polaires

Exemple *HF*

$$e_H \prec e_F$$

Molécules AB (fortement) polaires

Exemple *HF*

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$$E(1s_H) = -Ry = -0.5 \text{ u.a.}$$

$$1 \text{ u.a.} = 27.2 \text{ eV}$$

Molécules AB (fortement) polaires

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$$E(1s_F) \cong -38 \text{ u.a.}$$

$$E(2s_F) \cong -1.44 \text{ u.a.}$$

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$$E(1s_F) < E(2s_F) \ll E(1s_H) \cong E(2p_F)$$

Molécules AB (fortement) polaires

Exemple *HF*

$$e_H \prec e_F$$

$$\psi_1 \cong 1s_F$$

OM de coeur

$$1 \text{ u.a.} = 27.2 \text{ eV}$$

Molécules AB (fortement) polaires

Exemple *HF*

$$e_H \prec e_F$$

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OM de coeur **1 u.a. = 27.2 eV**

$$\psi_2 \cong 2s_F$$

OM non-liante

Molécules AB (fortement) polaires

Exemple *HF*

$$e_H \prec e_F$$

$$\psi_1 \cong 1s_F$$

OM de coeur **1 u.a. = 27.2 eV**

$$\psi_2 \cong 2s_F$$

OM non-liante

$$\psi_3 \cong 2p_{z,F} + t1s_H \quad 0 < t < 1 \quad \text{OM liante}$$

Molécules AB (fortement) polaires

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$$\psi_4 \cong 2p_{x,F} \quad \psi_5 \cong 2p_{y,F} \quad \text{OM non-liante}$$

Molécules AB (fortement) polaires

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$$\psi_1 \cong 1s_F$$

OM de coeur **1 u.a. = 27.2 eV**

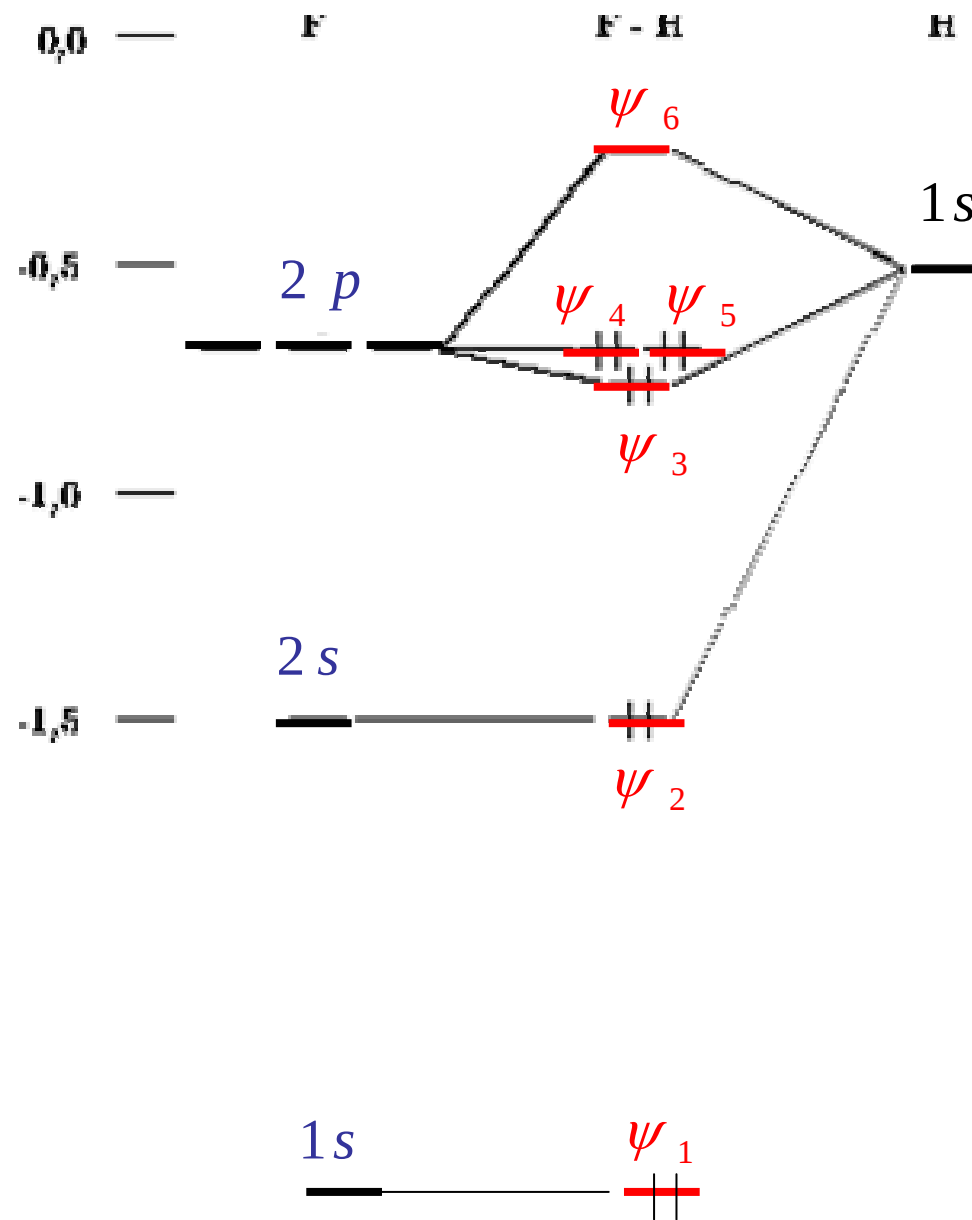
$$\psi_2 \cong 2s_F$$

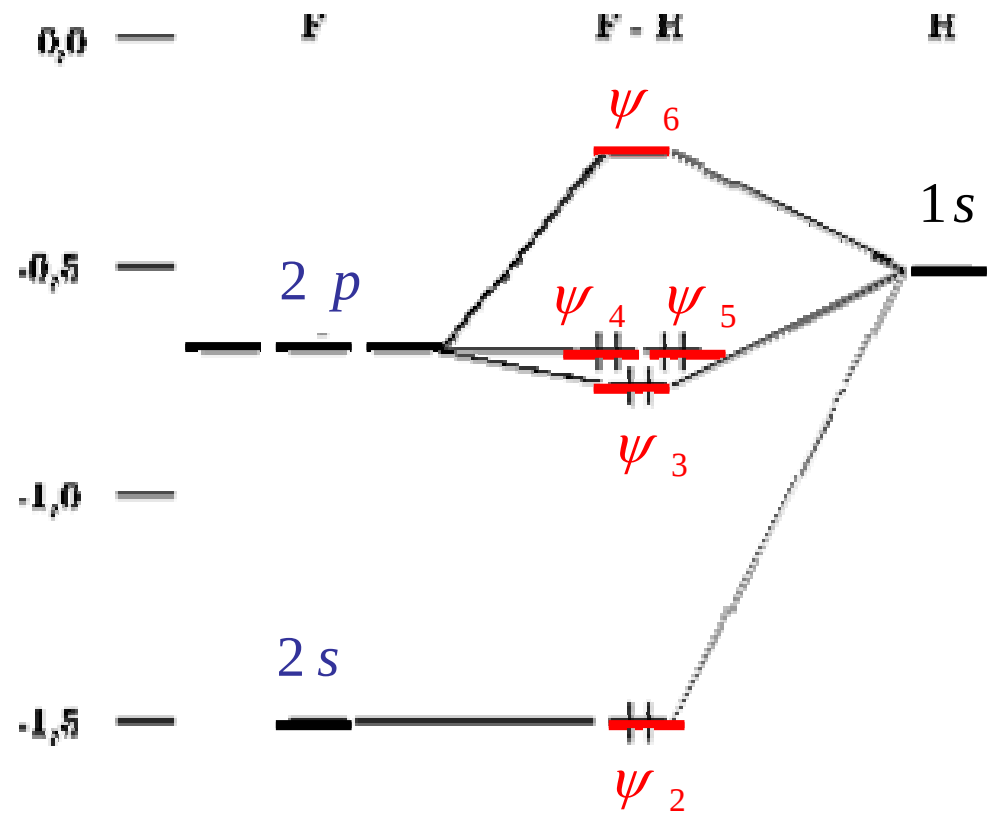
OM non-liante

$$\psi_3 \cong 2p_{z,F} + t1s_H \quad 0 < t < 1 \quad \text{OM liante}$$

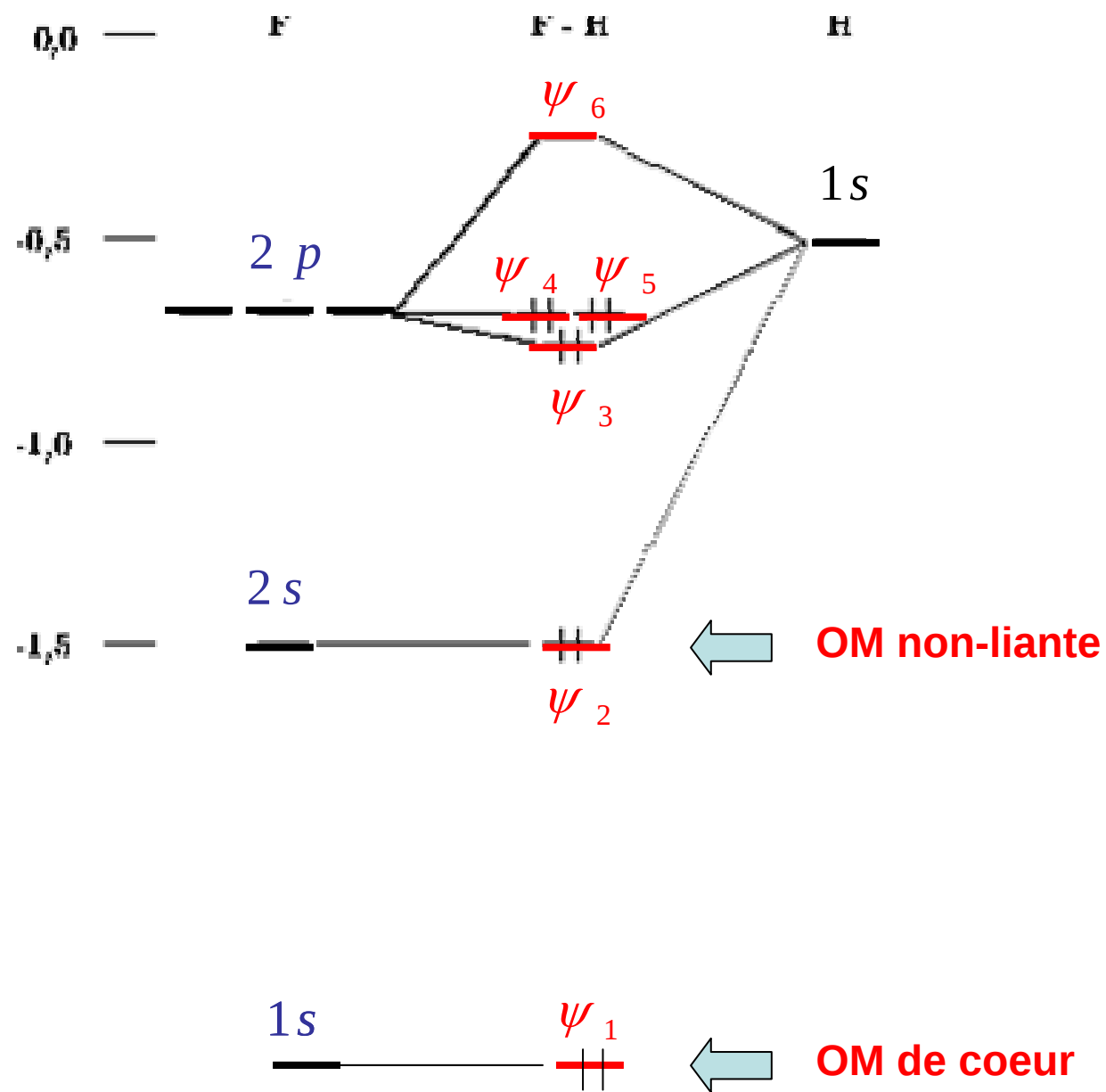
$$\psi_4 \cong 2p_{x,F} \quad \psi_5 \cong 2p_{y,F} \quad \text{OM non-liante}$$

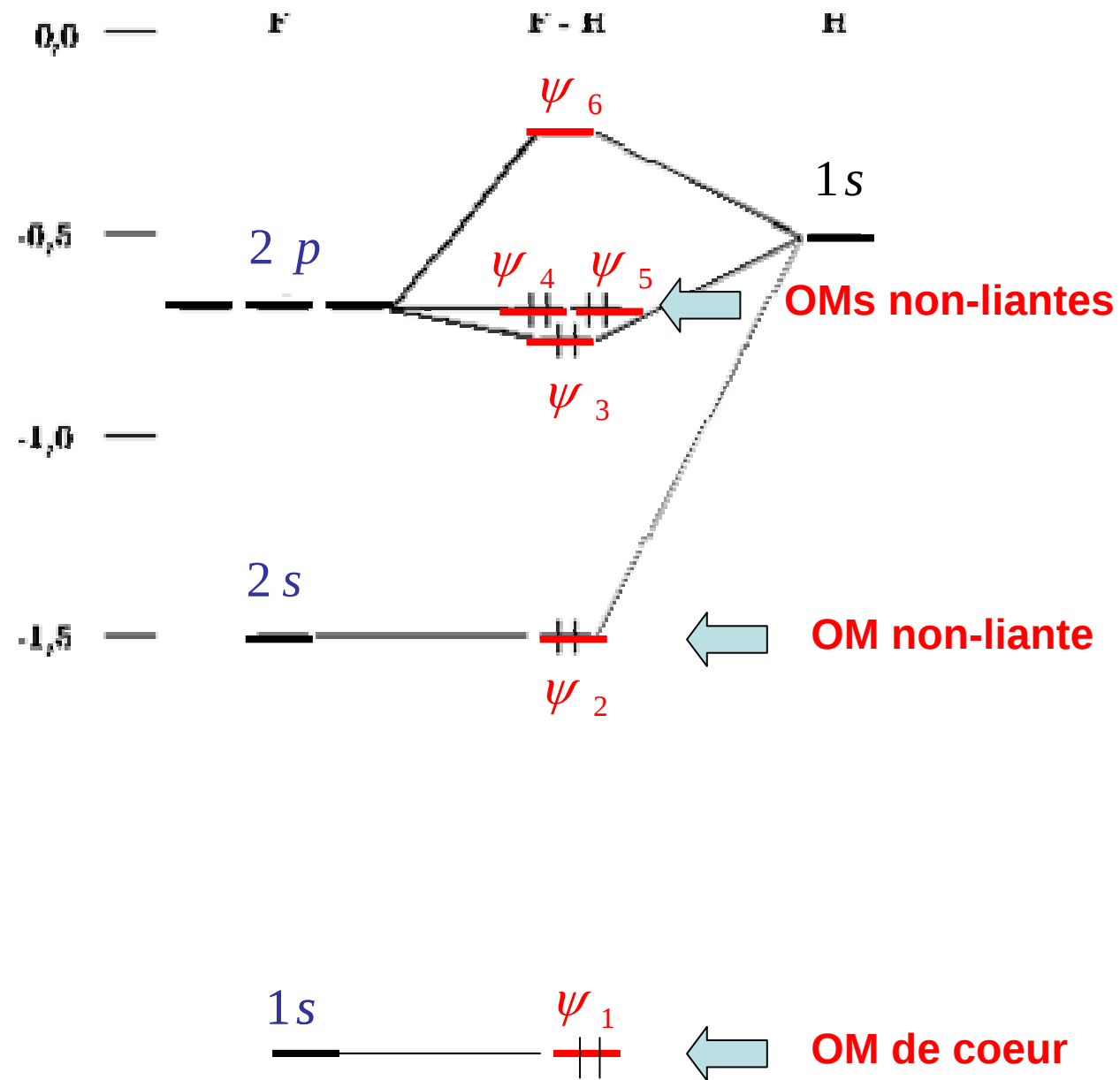
$$\psi_6 \cong 1s_H - t2p_{z,F} \quad 0 < t < 1 \quad \text{OM anti-liante}$$

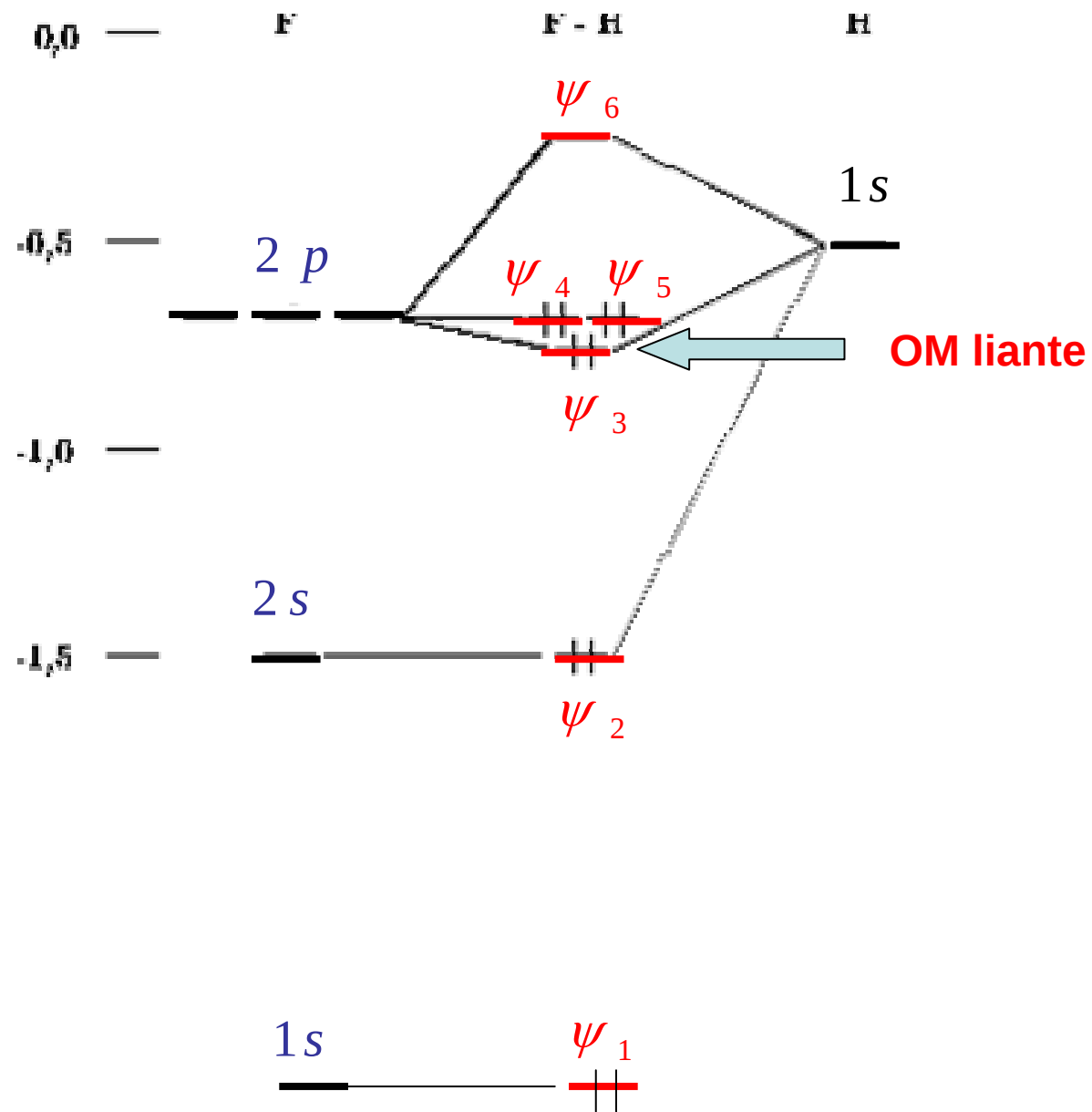


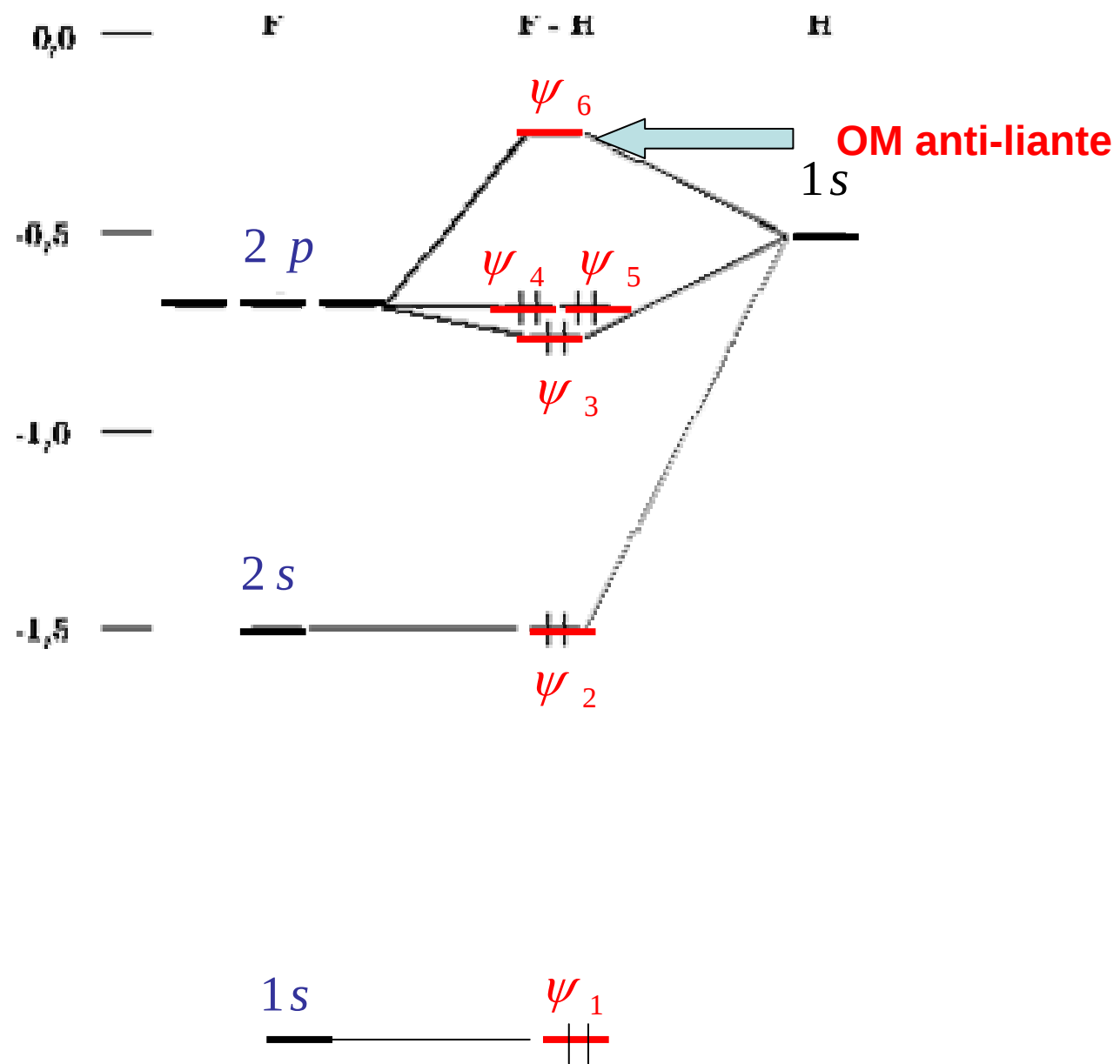


1s ψ_1 ← OM de coeur





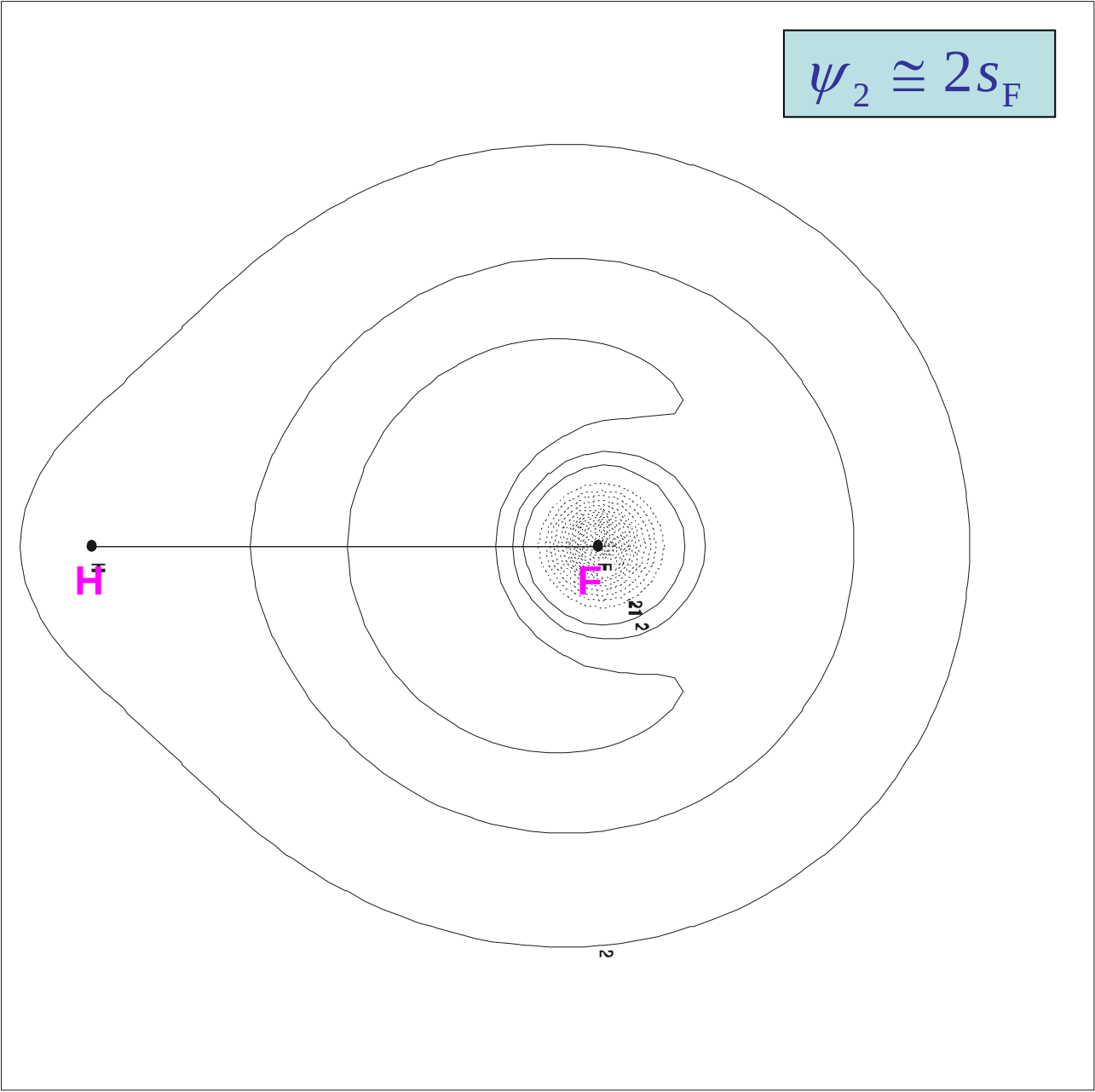




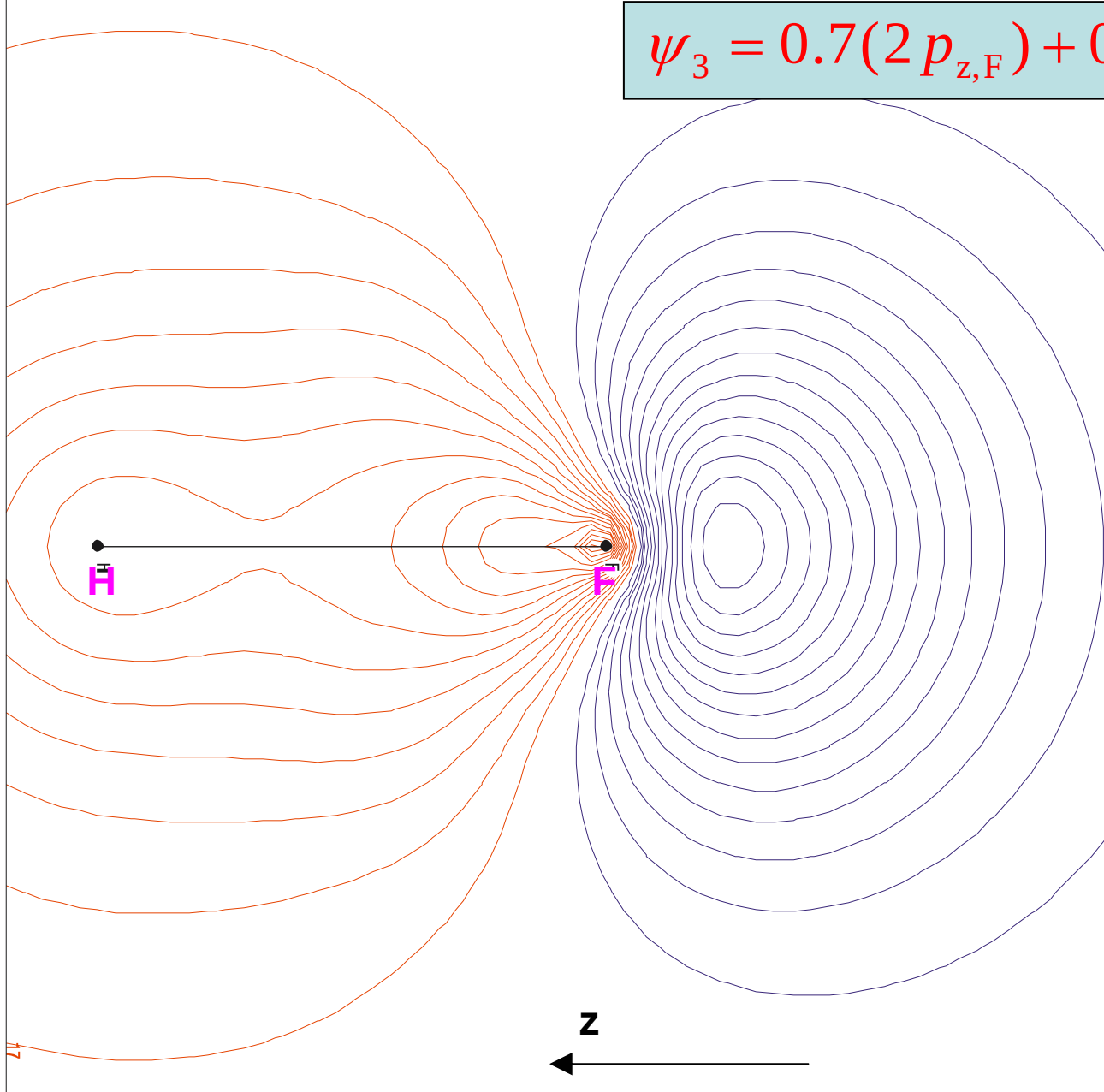
$$\psi_1 \cong 1s_F$$



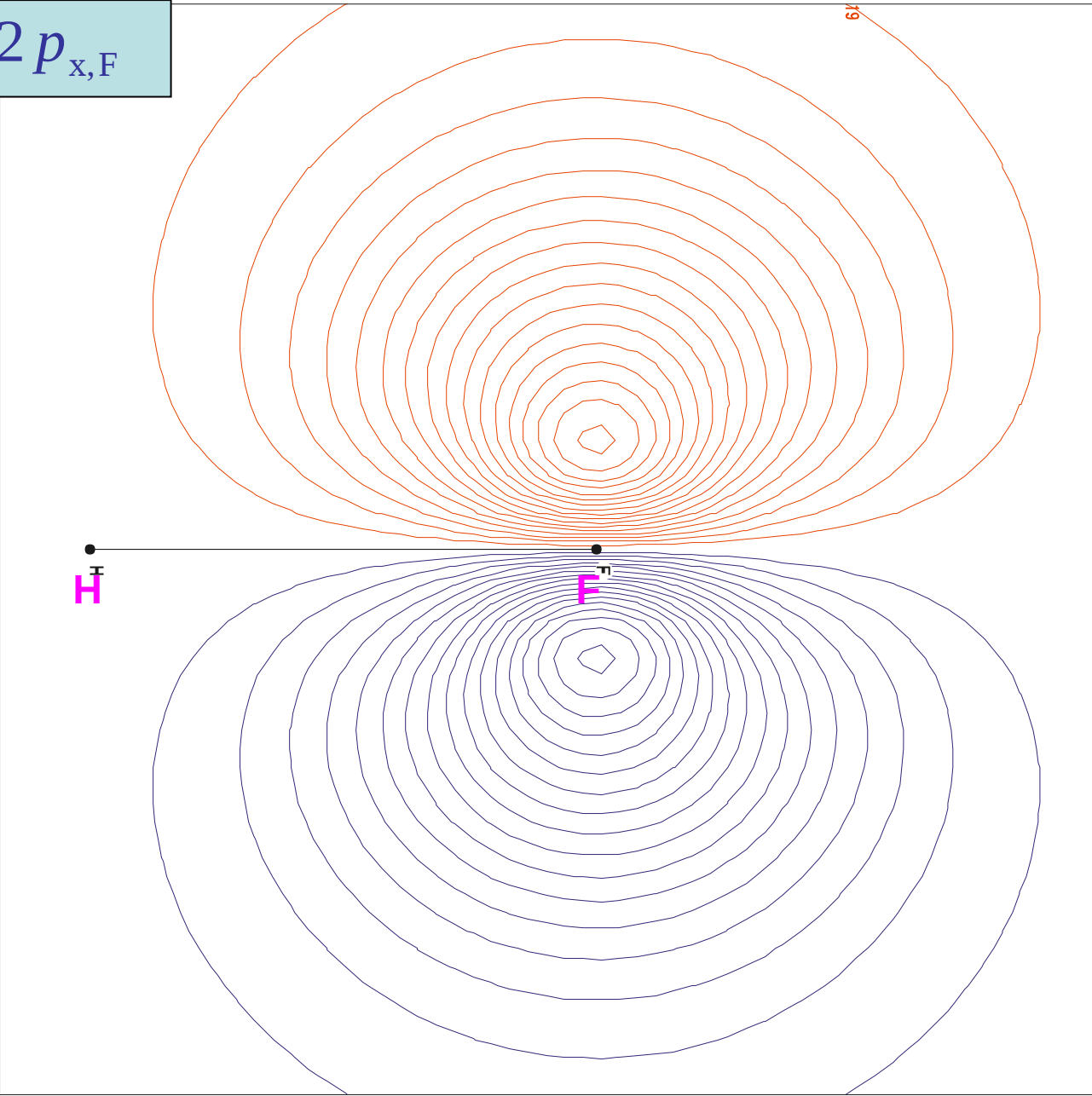
$$\psi_2 \cong 2s_F$$



$$\psi_3 = 0.7(2 p_{z,F}) + 0.5(1s_H)$$



$$\psi_4 \cong 2p_{x,F}$$



$$\psi_6 \cong 1s_H - 0.8(2p_{z,F})$$

